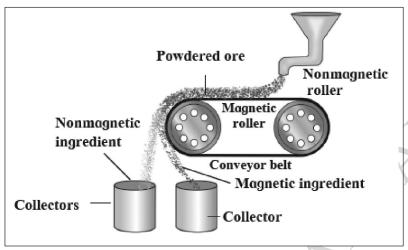


Subject : Science -I		Question Paper	Total Marks: 20
Class	: X	Topic: 8. Metallurgy	Time: 1 Hour
Q.1:	A) Choose the correct alto	ernative :	2
	1) The earthy impurities	associated with mineral used in metal	lurgy are called.
	a) Slag	b) Flux	
	c) Gangue	d) Mineral	
	2) A process employed for	r the concentration of sulphide ore is.	
	a) Froth flotation	b) Roasting	/
	c) Electrolysis	d) Bessemerisation	
	B) Solve the following qu	nestion. (Any One)	1
	1) Write name and molecula	ar formula of the common ore of aluminium	n.
	2) Give correlation		
	Bronze: Copper ar	nd Tin : : Stainless steel :	
	3) State true or false.	257	
	The melting and bo	iling points of ionic compounds are low.	
Q. 2:	A) Give scientific reason.	(Any One)	2
	1) Cryolite is added to alum	nina in the electolysis.	
	2) Calcium floats over water	er during the reaction with water.	
E	B) Solve the following que	stion. (Any Two)	4
	1) State four properties of i	onic compounds.	
	2) Write and explain the rea	action when steam is passed over aluminiu	m.
	3) Define: i) Calcination ii	ii) Galvanising	
	4) Why sodium is always k		
0.36	Solve the following question		6
2.00		n, Ca, Mg, Fe, Na, Li into three groups na	
	,		mery reactive
	,	ve metals and less reactive metals.	a o a 1:1 o1
	· ·	ation of metal 'A' is 2, 8, 1 that of metal 'E tive? write their reaction with dilute hydro-	

- 3) Explain any three chemical properties of metals with examples and reaction.
- 4) Explain Corrosion of metals and methods of preventing corrosion.

Q. 4: Solve the following questions. (Any One)

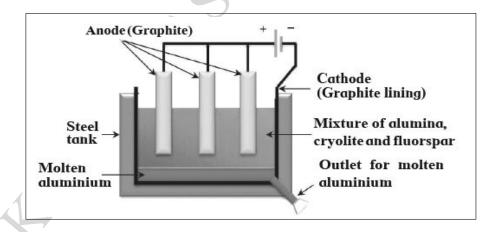
1) Answer the following question with reference to diagram.



5

Magnetic Separation

- i) Which machine is used in the magnetic seperation method?
- ii) Name the two parts of electromagnetic machine.
- iii) Name the magnetic and nonmagnetic ingredients of tin ore.
- iv) From which conveyor belt is made up of?
- v) Name of ore of tin.
- 2) Answer the following question with reference to diagram.



- i) Which process of extraction shown in figure?
- ii) Write cathode reaction.
- iii) Why cryolite is mixed with molten alumina?
- iv) Write the molecular formula of cryolite and flurospar.
- v) Which gas is liberate at anode?

* * *

