

Subje	et : S	Science -I	Question Paper	Total Marks : 20	
Class	: 2	X	5. Heat	Time: 1 Hour	
Q.1 A : Choose the correct alternative :					
1)	The	amount of water vapour	in air is determined in terms of its		
	a)	Humidity	b) Rancidity		
	c)	Fluidity	d) None of the above.	,	
2)	Dur	ing transformation of liqu	aid phase to solid phase the latent heat is		
	a)	Solidification	b) Fusion.		
	c)	Vaporization	d) None of the above		
В.	Solv	ve the following questio	n. (Any One).	1	
	1)	State whether the follow	ing statement is true or false		
		During transition of solid	d phase to liquid temperature of object increa	ases.	
	2)	Complete the analogy	5		
		SI unit of heat :	:: CGS unit of heat :		
	3)	Define one cal heat.	<i>Y</i>		
Q. 2 : A)	Giv	e scientific reason. (An	y One)	2	
1)	Aqu 0°C	= /	an survive even when atmospheric temperatu	are goes below	
2)	Dur	ing winter season a white	trail at back of flying plane is observed in a	clear sky.	
Q. 2 : B)	Solv	e the following questio	ns. (Any Two)	4	
1)	Hov 100		cessary to raise the temperature of 5kg of v	vater from 20°C to	
2)	Wri	te down the units of speci	fic heat capacity.		
3)	Wha	at is relative humidity & %	relative humidity?		
4)	Wha	at is the dew point temper	rature?		

Q. 3: Solve the following questions. (Any Two)

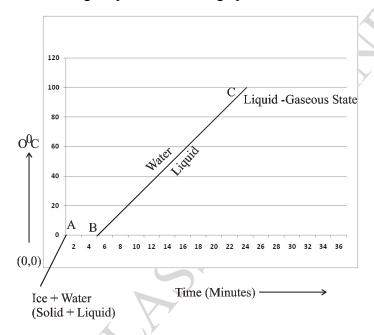
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- 1) Explain regelation.
- 2) Explain the role of latent heat in change of state of a substance.
- 3) A Calorimeter has mass 100 g and specific heat 0.1 kcal/kg⁰C. It contains 250 g of liquid at 30⁰C having specific heat of 0.4 kcal/kg⁰C. If we drop a piece of ice of mass 10g at 0⁰C what will be the temperature of mixture?
- 4) Which principle is used to measure specific heat capacity of a substance?

Q. 4: Solve the following question. (Any One)

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1) Explain the following temperature v/s time graph.



2) What is meant by specific heat capacity? How will you prove experimentally that different substances have different specific heat capacities?

