

Question Paper Subject: Chemistry

13: Nuclear Chemistry and Radioactivity Time: 1 Hour Class : XI

SECTION A

0.1 : Choose the correct option :

- i) The radioactive isotope used in the treatment of Leukemia is
 - a) 60C
- b) ²²⁶Ra
- c) ³²P
- d) ¹³¹I
- ii) The process by which nuclei having low masses are united to form nuclei with large masses is
 - a) chemical reaction
- b) nuclear fission
- c) nuclear fusion
- d) chain reaction
- iii) The process during which there is no change in atomic number of parents nucleus
 - a) alpha emission

b)

beta emission

- c) positron emission gamma emission
- d)

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- iv) The correct order of penetrating power is
 - a) $\gamma >> \beta > \alpha$

- c) $\alpha > \beta > \gamma$ d) $\alpha = \beta > \gamma$

Q.2 : Answer the following:

- i) What is nuclear chemistry?
- ii) Define: mess defect

SECTION B

- Answer the following: (ANY 2) 4
- : What is nuclear potential. What are its Q.3

consequences?

- Q.3 : Define half life of radio element. Derive an expression for half life of radioelement?
- Q.4 : Explain γ decay process in radioactivity.

SECTION C

: Answer the following : (ANY 2)

Total Marks:20

- The half life of ^{20q}P₀ is 102y. How much of Q.5 : 1 mg sample of polonium decays in 62y?
- Derive an expression for decay constant for Q.6 : a radioactive element.
- Q.7: Derive an expression for nuclear binding energy.

SECTION D

- Answer the following: (ANY 1)
- Q.8 How are nuclides classified on the basis of number of neutrons and protons?
- Q.9 Define the following:
 - i) Radio activity
 - ii) Radioactive elements.
 - iii) Name the radioations emitted by radioactive elements.
 - iv) What is rate of radioactive decay?

